SESQUITERPENE LACTONES OF CALEA ZACATECHICHI AND C. URTICIFOLIA*

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Key Word Index—Calea zacatechichi; C. urticifolia; Compositae; Heliantheae; Galinsoginae; sesquiterpene lactones; heliangolides; flavones.

Abstract—Calea zacatechichi yielded the sesquiterpene lactone zexbrevin and a new analog, several analogs of neurolenin B including calein A, two analogs of budlein A and the flavones acacetin and O-methylacacetin. Calea urticifolia contained additional analogs of neurolenin B as well as a series of epoxidation products. Structures were established by spectrometric techniques. The results are contrasted with previous findings.

INTRODUCTION

Two recent articles [1, 2] dealing with the constituents of Calea zacatechichi reported isolation of different sesquiterpene lactones. Moreover, the spectroscopic data reported for the caleins from C. zacatechichi [2] and for several apparently similar lactones isolated from C. urticifolia [3] seemed more in keeping with structures of type 2 analogous to neurolenin A (2a) and neurolenin B (2b) from Neurolaena lobata [4] than with the formulae [2] or the stereochemistry [3] originally assigned to them. We have therefore reinvestigated these two species. Our own work on collec-

1a R = MeAcr **1b** R = Tigl

tions from Honduras has resulted in isolation of the sesquiterpene lactones 1a (zexbrevin), 1b, 2c-2e, 3a and 3b from C. zacatechichi, whereas C. urticifolia gave 2f-2h as well as 4a-4e. Substance 2e was identical with calein A [2] whose structure must be revised.† Compounds 2c, 2f and 2g have been found previously in C. urticifolia but were assigned different stereochemistry [3].

2a R = O - iVal, R' = H

2b R = O - iVal, R' = OAc

2c R = OMeAcr, R' = OAc or reverse

2d R = OTigl, R'' = OAc or reverse

2e R = OAng, R' = OAc or reverse

2f R = OAng, R' = OMcAcr or reverse

2g R, R' = OMeAcr

2h R = OiBu, R' = OMeAcr or reverse

3a R = Tigl, R' = OH

3b R = MeAcr, R' = OH

3c R = Ang, R' = OH

4a R = MeAcr, R' = Ang or reverse

4b R = MeAcr, R' = iVal or reverse

4c R, R' = MeAcr

4d R = MeAcr, R' = iBu or reverse

4e R = MeAcr, R' = Ac or reverse

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[†]Dr. A. Romo de Vivar has informed us (March 1979) that he has also come to the conclusion that calein A and calein B possess structure **2e** and its reverse; see also Quijano, L., Romo de Vivar, A. and Rios, T. (1979) *Phytochemistry* **18**, 1745.

RESULTS AND CONCLUSIONS

The pair of lactones **1a** and **1b** could not be separated satisfactorily; analysis of the 270 MHz ¹H HMR and 67.09 MHz ¹³C NMR spectra by spin-decoupling and single frequency off-resonance decoupling when feasible permitted the assignments given in Tables 1 and 2. Lactone **1a** appeared to be identical with zexbrevin [5] to which we have recently [6] assigned H-8 α rather than H-8 β stereochemistry on the basis of a correlation with tirotundin.

Lactone pairs **2c** and **2d**, on the one hand, and **2g** and **2h** on the other, could not be separated satisfactorily, but analysis of the ¹H and ¹³C NMR spectra and solvent shift studies permitted identification of the various constituents. Extensive spin decoupling in different solvents and comparison (see Tables 1 and 2)

with the ¹H and ¹³C NMR spectra of neurolenin B (**2b**) whose structure was established by X-ray crystallography [4] clearly showed that lactones **2c-2h** had structures and stereochemistries entirely analogous to those of the neurolenins. The remaining uncertainty as to the distribution of ester side chains between C-8 and C-9 could not be resolved since selective hydrolysis of one of the ester functions was not successful.

Lactone **2e** was identical with calein A from Mexican C. zacatechichi [2] by direct comparison; the structures of this substance and of calein B therefore require revision. Lactones **2c**, **2f** and **2g**, on the other hand, appeared to be identical with those lactones which were reported earlier [3] from a Guatemalan collection of C. urticifolia, but were assigned enantiomeric stereochemistry at C-4 on the basis of paramagnetic shifts of H-4 and H-7 during NaBH₄

Table 1. ¹H NMR spectra of Calea constituents*

Compound	H-2	H-3	H-4	H-5	H-6	H-7	H-8	H-9	H-13	H-14†	H-15†	Misc.
1a,b	5.54br		3.03quibr	2.59ddd	4.48ddbr	3.30m	5.13ddd	2.70dd	6.34d	1.4	1.39d	‡, §
	(1)		(7, 1)	(15, 9, 7)	(9, 5, 0.d)	(5, 2.3, 1)	(5, 2.5, 1)	(15.5, 5)	(3.5)		(7)	
				2.05ddbr				2.23dd	5.68d			
								(15.5, 2.5)	(3)			
2a	6.58d	5.881	3.08 <i>tqui</i>	1.81td	4.50dd	2.61br	5.29ddd	2.25dd	6.27br	1.44	1.13d	ļi.
	(12)	(12)	(12, 6)	(12, 5.5)	(12, 5.5)	(3, 1.5, 1)	(11, 6, 3,)	(14.5, 11)	(1)		(6)	
				1.42td				2.03dd	5.73d			
				(12, 6)				(14.5, 6)	(1.5)			
2b	6.58d	6.00t	3.10tqui	1.83td	4.54dd	2.5qm	5.5m	5.5m	6.32br	1.34	1.14d	2.05 (Ac)
	(12)	(12)		1.43td					5.8br			ji.
2b††	5.99d	4.931	2.79tqui	1.24td	4.34dd	2.43m	5.91dd	5.64d	6.18d	1.09	0.58d	
				0.95td			(10, 1.5)	(10)	5.20d			
2c,d	6.60d	6.01t	3.10 tqui	1.81td	4.60dd	2.63br	5.60br	5.60br	6.31br	1.34	1.13d	2.03 (Ac)
				1.42td					5.82br			1.97 (Ac)
2¢††	6.02	5.00t	2.80 tqui	1.26td	4,40dd	2.48br	5.98dd	5.69d	6.17br	1.11	0.61d	1.64 (Ac)
			-	0.99td					5.27br			
2d	6.06d	5.08d	2.80tqui	1.26td	4.48dd	2.48br	5.98dd	5.73d	6.17br	1.15	0.56d	1.64 (Ac)
				1.00td					5.31br			8
2e	6.60d	6.02t	3.12tqui	1.84td	4.58dd	2.64br	5.68dd	5.59d	6.33br	1.37	1.14	2.00 (Ac)
			•	1.44					5.84br			•
2f	6.61d	6.00t	3.10taui	1.82td	4.62dd	2.63br	5.61dd	5.74d	6.29br	1.33	1.14d	‡, €
				1.46td					5.81br			
2g	6.63d	6.01r	3.10tqui	1.85td	4.61dd	2.64dd	5.67br	5.67br	6.32br	1.34	1.14d	¶ (double)
- 0				1.45td					5.82br			
2h	6.61d	6.00t	3.10tqui	1.85td	4.61dd	2.63br	5.64br	5.64br	6.31br	1.33	1.13d	į, **
				1.45td					5.82br			
3a	5.68	_		6.23dr	5.31di	3.75m	5.19ddd	2,54dd	6.34d	1.49	4.41br**	š
	-			(4, 1.5)	(4, 1)	(4, 3.3, 3, 1.5)	(5.5, 3.5, 1.5)	(14.5, 5.5)	(3.3)		(1.5, 1)	ţ.
						(,, = ,= , , , = ,= ,		2.31dd	5.68d			
								(14.5, 3.5)	(3)			
3b	5.70	_		6.22dt	5.30dt	3.77	5.20	2.60dd	6.35d	1.49	4.42br**	‡
				3. _	0.00	2		2.32dd	5.69d	****		·
4a,b	4.29d	3.33dd	1.47m	1.90m	4.84dd	2.37br	5.72dd	5.89d	6.32br	1.45	1.22d	‡, ¶ and ‡,
7,	(4.5)	(9, 4.5)	1.4,	1.48	(12, 5.5)	(1.5, 1, 1)	(10, 1.5)	(10)	(1)	1.70	(6)	3.90 (OH)
	(1.5)	(2), 1.27		2.70	(12,0.0)	13.5, 3, 4,	(10, 175)	(****	5.82br		(0)	5.50 (011)
									(1)			
4e	4,29d	3.33dd	1.46m	1.90m	4.82dd	2.83br	5.83br	5.83br	6.32br	1.46	1.23d	‡ (double)
	.,274	0.0044	2.1011	1.48	1.0200	210301	2.030	5,050	5.82br	1.70	1.2.74	3.91 (OH)
4d	4.25d	3.32dd	1.46m	1.46 1.94m	4.82dd	2.36br	5.70dd	5.78dd	6.32br	1.45	1.24d	\$, **
4e		J.J244		1.48	7.02uu	2.5007	5.7044	J. 1000	0.0001	1.77.7	1.270	3.89 (OH)
	4.23d	3.33dd	1.47m	1.40 1.97m	4.80dd	2.35br	5.66dd	5.76d	6.32 <i>br</i>	1.47	1.22d	2.04 (Ac)
	1,400	5,5544	A . TF / 116	1.47	7.0000	2.5501	J.00uu	J. 10a	5.82br	1.47	1.2214	3.94 (OH)

^{*} Run at 270 MHz in CDCl₃ unless indicated otherwise with TMS as internal standard. Values are in ppm. Unmarked signals are singlets. Figures in parentheses are coupling constants in hertz and are not repeated in the same column for similar compounds.

[†] Intensity three protons.

[‡] MeAcr—1.87dd (1.5, 1, H-3')†, 5.97dq (2, 1, H-4'), 5.59dq (2, 1.5, H-4').

[§] Tigl—6.72qbr (7, 1, H-3'), 1.79 (7, H-4'),† 1.75br (1, H-5').†

^{||}i-Va|-2.25m, 2.03m (H-2'), 1.99m (H-3')‡‡, 0.88d(6H, H-4').

[¶] Ang—6.10qq (7.5, 1.5, H-3'), 1.94dd (7.5, 1.5, H-4'),† 1.76q (1.5, H-5').†

^{**} i-Bu-2.55sept (7.5, H-2'), 1.07d (6H, H-3').

^{††} In C₆D₆.

^{‡‡} Intensity two protons.

Table 2. 13C NMR spectra of Calea constituents*

Carbon	1a (1b)	2a †	2b†	2c (2d)	2e	2f	2g (2h)	3a	3b	4a (4b)	4c (4d)	4e
1	205.18	205.7	204.3	204.41	204.60	204.85	204.71	205.12	205.42	206.07 (206.75)	205.98	205.72
2	102.95	125.3d	125.3d	125.58d	125.39d	125.58d	125.48d	104.94d	104.94b	55.88d	55.94d	55.70d
3	192.13	146.6d	147.9d	147.92d	148.07d		147.92d	182.76	182.76	62.81d	62.83d	62.88d
4	31.37d	28.3d	28.2d	28.21d	28.29d	28.29d	28.29d	138.88	138.88	26.08d	25.14d	25.96d
5	40.96t	40.2t	40.21	40.17t	40.36t	40.30t	40.28t	133.85d	133.85d	38.92t	38.99t	38.72d
6	74.73d	75.9d	76.3d	76.35d	76.31d	76.37d	76.35d	75.25d	75.28d	74.79d (74.16d)	74.81d	74.73d
7	51.61d	42.4d	41.2d	41.18d	41.25	41.29d	41.23 <i>d</i> (41.34 <i>d</i>)	48.25d	48.23d	41.12d	41.12d (41.23d)	40.83 <i>b</i>
8	74.25d (74.41d)	73.9d	73.9d	74.52d‡ (74.11d)‡	;	74.46d‡	74.30d‡			74.16d	74.11d	74.05d
9	43.06t (43.10t)	40.4d	73.8d	74.00d‡	74.05d‡	73.07d‡	74.05d‡ (73.23d)		42.631	70,89d (71,33d)	72.01d (71.20d)	71.68 <i>d</i>
10	88.29	76.5	79.3	79.28	79.28	79.34	79.40	88.00	87.99	79.93	80.04	79.74
11	139.53 (139.64)	136.1	134.8	134.77§	134.82	134.87§	134.85§	136.00	135.99	134.98‡	134.76‡	134.33‡
12	168.46 (168.54)	169.0	170.0	170.15	170.14	168.70	168.67	168.95	168.95	168.34	168.29	170.34
13	123.05t (122.91t)	124.5 <i>t</i>	126.21	126.251	126.26t		126.33 <i>t</i> (126.49 <i>t</i>)	123.381		126.35t	126.35 <i>t</i> (126.48 <i>t</i>)	126.67 <i>t</i>
14	22.74q	28.1q	_	23.50q	23.67q	23.83q	23.63q	-	21.38q	24.75q	24.55q	24.43q
15	16.04	19.9q	19.6q	19.64q	19.71q	19.72q	19.72q	62.30t	62.26t	18.57q	18.57q	18.52q
1′	165.44 (166.12)	171.3	170.8	165.19 <i>d</i> (165.78)	165.21	165.19	165.32 (165.19)	165.40	166.44	165.24	165.10 (165.05)	165.21
2′	135.12 (127.50)	43.0 <i>t</i>	42.5 <i>t</i>	134.90§ (127.25)	126.51	134.71§	134.63§	134.51	127.33	134.63‡	134.71‡	134.79‡
3'	18.05q (138.83d)	25.4d	24.9d	17.98q (138.83d)	140.79d	17.98q	18.06 <i>q</i> (17.98 <i>q</i>)	18.24q	139.76d	17.92q	17.97q (17.98q)	18.03q
4'	126.70 <i>t</i> (14.54 <i>q</i>)	22.2q	22.3q	126.85t (14.46q)		126.59t	127.60t (127.19t)	127.35t	14.62 <i>q</i>	126.62t (127.01t)	127.84t (127.13t)	127.84t
5′	(11.90q)	22.2q	22.3q	(11.76q)	15.71q				11.84q			
1"			168.6	168.66 (168.73)	168.72	166.58	166.50 (176.50)			166.77 (173.00)	166.74 (176.50)	168.37
2"			20.5q	20.31q (20.31q)	20.31	126.46	134.69 (33.96 <i>d</i>)			126.43 (42.45t)	126.35 (34.07)	20.26q
3"						141.14d	18.00q (18.65q)			141.65d (24.56d)	17.97q (16.57q)	
4"						20.12q	126.73 <i>t</i> (19.01 <i>q</i>)			20.00q $(15.15q)$	126.70 <i>t</i> (18.95 <i>q</i>)	
5"						15.85q	` '			15.80q (10.14q)	•	

^{*}Run at 67.09 MHz in CDCl₃ on a Bruker HX-270 Instrument with TMS as internal standard. Values are in ppm. Unmarked signals are singlets. Values in parentheses correspond to those signals of pair which differ in shift.

reduction of 2c. However, the coincidence of chemical shifts, coupling constants and other physical properties leaves no doubt that the stereochemistry of these compounds is as shown in formulae 2c-2h.* The observed coupling constants adduced in support of the earlier stereochemistry [3] are equally compatible with the conformation of the neurolenins established by X-ray crystallography [4]. Inspection of the models and of Fig. 2 of ref. [4] shows that the ketone carbonyl is not in the plane of the 2,3-double bond; this accounts not only for the frequency of the C-1 signal in

the 13 C NMR spectra of **2a–2h** (\sim 205 ppm) which is characteristic of a saturated, not an α,β -unsaturated cyclohexenone carbonyl, but also, in the 1 H NMR spectra, for an interesting inversion in the frequencies of the α - and β -protons. The signal of H-2 appeared as a doublet near δ 6.6 and that of H-3 as a triplet near 6 ppm (not a doublet as reported in ref. [4]). The unusual paramagnetic shift of H-2 which lies fairly close to the x-axis at the 'carbon end' of the carbonyl group is difficult to explain in terms of the usual model, but may be related to similar observations discussed elsewhere [7]. On the other hand, the IR band near $1690 \, \mathrm{cm}^{-1}$ which is that of the ketone carbonyl (not that near $1735 \, \mathrm{cm}^{-1}$ as claimed earlier [3] which is that of the saturated ester function of

[†] Taken from ref. [4]. Assignments of C-4 and C-3', and C-12 and C-1', have been interchanged.

^{‡, §} Assignments interchangeable.

Probably misprints.

^{*}In the case of our **2c** from *C. urticifolia*, a difference in the distribution of the acetate and methacrylate esters between C-8 and C-9 cannot be excluded with certainty.

2a–2f and **2h**) indicates a certain degree of conjugation (or possibly hydrogen bonding) as is evident by comparison with the IR spectra of **4a–4e** (vide infra).

Lactones **4a-4e** from C. urticifolia were clearly α ketoepoxides and biological oxidation products of lactones of type 2. This was evidenced by the shift of the ketone IR frequency to longer wavelengths (near 1705 cm⁻¹), and by the diamagnetic shift of the signals of H-2 and H-3 to near δ 4.4 and 3.3, respectively. Corresponding upfield shifts of C-2 and C-3 were noted in the ¹³C NMR spectra, while the frequency of C-1, out of the plane of the former double bond, remained invariant. Spin decoupling established the complete sequence of protons on C-2 through C-10. The stereochemistry assigned to the new asymmetric centres was based on coupling constants involving H-2, H-3 and H-4 and, in the rest of the molecule, on the correspondence between the NMR spectra of lactones of type 2 and type 4. Lactones 4a,b and 4c,d could not be separated satisfactorily and, as in the case of the lactones of type 2, the allocation of the two ester functions to C-8 and C-9 of **4a,b,d** and **e** remains uncertain for the time being.

The gross structure of lactones 3a,b from our collection of C. zacatechichi was established by the IR and UV spectra, double resonance experiments and the 13 C NMR spectrum (Tables 1 and 2). That they were analogs of budlein A (3c) [8] from Viguiera buddleiaeformis rather than goyazensolide (3b with a trans-lactone ring closed to C-8) [9] was clear from the chemical shifts of H-8 (near δ 5.3 instead of 4.5) and the coupling constants which defined the stereochemistry at C-8. Derivatives of 3 carrying an additional hydroxyl group on C-10 and 15-desoxy analogs with the same stereochemistry have been isolated from C. urticifolia [3] and C. axillaris [10, 11] and from Isocarpha atriplicifolia [12], Helianthus ciliaris [10] and H. lehmannii [13].

There are obvious similarities in the sesquiterpene lactone chemistry of Calea (Heliantheae subtribe Galinsoginae) and several representatives of subtribe Helianthinae. The very limited results based on examination of a mere handful of species also suggest similarities in the sesquiterpene lactone chemistry of Calea and Neurolaena, a genus recently transferred from Senecioneae to Heliantheae and placed in a newly erected subtribe Neurolaeninae which is close to Galinsoginae [14]. Additional chemical studies may shed further light on these relationships.

EXPERIMENTAL

Extraction of C. zacatechichi. Aerial parts of C. zacatechichi Schlecht. (12 kg), collected by Mr. Gustavo Cruz in the fall of 1975 in the vicinity of Tegucigalcapa, Honduras (voucher on deposit in herbarium of Universidad Nacional Autonoma de Honduras, coll. No. 4116 and accession No. PR-80864 of Medicinal Plant Resources Laboratory, U.S.D.A.), were extracted with CHCl₃ and worked up in the usual manner [15]. The crude gum (75 g) was adsorbed on 500 g of Si gel (Mallinckrodt 100 mesh) and chromatographed over 1 kg of Si gel (toluene–CHCl₃, 4:1), 500 ml fractions being eluted in the following order: 1–5 (toluene–CHCl₃, 4:1), 6–11 (toluene–CHCl₃, 1:1), 12–26 (CHCl₃–MeOH, 99:1), 27–35 (CHCl₃–MeOH, 97:3), 36–45 (CHCl₃–MeOH, 19:1). TLC of fractions 6–11 showed a

major spot; PLC (EtOAc-hexane, 3:7) and crystallization from hexane gave material, mp 189-200°, which was homogeneous on TLC in several solvent systems but was a 5:1 mixture of 1a and 1b (¹H and ¹³C NMR and MS analysis, $[\alpha]_D + 60^\circ$ (c 0.03, CHCl₃); CD curve (MeOH) $[\theta]_{304} + 1390, [\theta]_{281} - 630, [\theta]_{250} + 26700, [\theta]_{218} - 2770$ (last reading); IR $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$: 1763, 1720, 1690, 1655, 1630 and 1592. (Calc. for $\mathbf{1a}$, $C_{19}H_{22}O_6$: MW, 346.1415. Found: MW(MS), 346.1427 (6.2%). Calc. for 1b, C₂₀H₂₄O₆: MW, 360.1571. Found: MW(MS), 360.1579 (10.8%)). Other significant peaks in the high resolution MS were at m/e (rel. int.): $316 (C_{18}H_{20}O_5, 2.4), 302 (C_{17}H_{18}O_5, 3.0), 261$ $(C_{15}H_{17}O_4, 12.2), 260 (C_{15}H_{16}O_4, 40), 233 (C_{12}H_{17}O_3, 3.7),$ $232 \ (C_{14}H_{16}O_3,\ 11.2),\ 217 \ (C_{13}H_{13}O_3,\ 7.4),\ 204 \ (C_{12}H_{12}O_3,$ 7.1), 125 ($C_7H_9O_2$, 100), 124 ($C_7H_8O_2$, 66.5), 83 (C_5H_7O , 83.7), 82 (C_5H_8O , 52.9), 69 (C_4H_5O , low resoln.). Fractions 12-20 gave a yellow solid which was recrystallized from CHCl₃-MeOH to give 7-O-methylacacetin, mp 158°. UV and NMR spectra as reported. This flavone has been isolated previously from C. zacatechichi [16]. Fractions 21-26 were combined and purified by PLC (EtOAc-hexane, 1:1 3 developments). The slightly more mobile component was recrystallized from CHCl3-hexane, mp 180-182°, yield 0.10 g, but was a 4:1 mixture of 2c and 2d (1H and 13C NMR spectra, MS, lit. [3] mp of **2c** 168°; $[\alpha]_D = 290^\circ$ (c 0.05, CHCl₃); CD curve (MeOH) $[\theta]_{302} - 350$, $[\theta]_{256} + 3000$, $[\theta]_{214} - 100\,500$ (last reading); IR $\nu_{\rm max}^{\rm KBr}\,{\rm cm}^{-1}$: 3420, 1770, 1745, 1730, 695, 1665, 1650, 1630 and 1230. (Calc. for 2a, $C_{22}H_{28}O_8$: MW, 420.1725. Found: MW(MS), 420.1757 (4.4%). Calc. for 2b. C₂₁H₂₆O₈: MW, 406.1626. Found: MW(MS), 406.1635 (0.6%)). Other significant peaks in the high resolution MS were at m/e (rel. int.): 378 ($C_{20}H_{26}O_7$, 1.5), 364 ($C_{19}H_{24}O_7$, 9.4), $346 (C_{19}H_{22}O_6, 9.4)$, $321 (C_{17}H_{21}O_6, 1.1)$, 320 $(C_{17}H_{20}O_6, 1.3), 279 (C_{15}H_{19}O_5, 5), 278 (C_{15}H_{18}O_5, 209),$ 277 ($C_{15}H_{17}O_5$, 13.9), 261 ($C_{15}H_{17}O_4$, 9.9), 260 ($C_{15}H_{16}O_4$, 12.8), 83 (C_5H_7O , 100), C_5H_6O (96.8), 69 (C_4H_5O , low resoln.). The slower moving component 2e was recrystallized from CHCl₃-hexane, mp 178°, yield 0.15 g, $[\alpha]_D$ = 344° (c 0.04, CHCl₃); CD curve (MeOH) $[\theta]_{302}$ – 2100, $[\theta]_{255}$ + 2000, $[\theta]_{210}$ - 6300 (last reading); MS m/e: 420 (M⁺), 400, 378, 360, 335, 278, 261, 83 and 60. Fractions 27-35 contained two very similar substances which were separated by PLC (Si gel, EtOAc-hexane, 1:1, 4 developments). The more mobile component **3a** was a gum, yield 70 mg, $[\alpha] - 80^{\circ}$ (c 0.03. CHCl₃), CD curve $[\theta]_{320} + 28000$, $[\theta]_{286} - 2800$, $[\theta]_{250} +$ 5700; IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3500, 1720,1690,1660,1640 and 1592. (Calc. for $C_{20}H_{22}O_7$: MW, 374.1365. Found: MW(MS), 374.1353 (12.9%)). Other prominent peaks in the high resolution MS were at m/e (rel. int.): 356 ($C_{20}H_{20}O_6$, 0.9), 330 $(C_{18}H_{18}O_6, 1.2), 291 (C_{15}H_{15}O_6, 1.6), 275 (C_{15}H_{15}O_5, 2.4),$ 274 ($C_{15}H_{14}O_5$, 5.6), 256 ($C_{15}H_{12}O_4$, 1.2), 248 ($C_{13}H_{12}O_5$, 12.3), 138 ($C_7H_8O_3$, 10.5), 83 (C_5H_7O , low resoln.). The less mobile component 3b was also non-crystalline, yield 60 mg, $[\alpha]_D - 87^\circ$ (c 0.023, CHCl₃); CD curve (MeOH) $[\theta]_{318} +$ 1080, $[\theta]_{288}$ - 1050, $[\theta]_{250}$ + 2160; IR $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$: 3500, 1765, 1720, 1680, 1640 and 1590. (Calc. for $C_{19}H_{20}O_7$: MW, 360.1208. Found: MW(MS), 360.1198 (16.4%)). Other significant peaks in the high resolution MS were at m/e (rel. int.): 342 (C₁₉H₁₈O₆, 1.4), 316 (C₁₇H₁₆O₆, 3.7), 291 $(C_{15}H_{15}O_6, 2.9), 275 (C_{15}H_{15}O_5, 3.5), 274 (C_{15}H_{14}O_5, 9.6),$ 248 ($C_{13}H_{12}O_5$, 37.4), 138 ($C_7H_8O_3$, 31), 69 (C_4H_5O , low resoln.). Fractions 36-42 gave a yellow solid which after recrystallization from MeOH-CHCl₃ melted at 262°, yield 0.12 g, and was identified as acacetin.

Extraction of C. urtificolia. Aerial parts of C. urtificolia (Mill.) DC (3.4 kg), collected by Mr. Gustavo Cruz in the fall

of 1974 in the vicinity of Tegucigalpa, Honduras (voucher on deposit in herbarium of U.N.A.H., accession No. PR-43871 of Medicinal Plant Resources Laboratory, U.S.D.A.), were extracted and worked up as usual. The crude gum (25 g) was adsorbed on 30 g Si gel and chromatographed over 300 g of the same adsorbent (toluene-CHCl3, 1:1), 250 ml fractions being eluted in the following order: 1-5 (toluene-CHCl₃, 1:1), 6-13 (CHCl₃), 14-20 (CHCl₃-MeOH, 99:1), 21-26 (CHCl₃-MeOH, 97:3), 27-34 (CHCl₃-MeOH, 19:1), 35-45 (CHCl₃-MeOH, 9:1). Fractions 6-13 contained a major component which was recrystallized (CHCl3-hexane) to give **2f**, mp 161–162°, lit. [3] mp 158.5°, yield 0.10 g, $[\alpha]_D = 400^\circ$ $(c \ 0.013, \text{ CHCl}_3); \text{ IR } \nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}; 3420, 1770, 1735, 1730,$ 1692, 1660, 1640 and 1625. Fractions 14-20 were further purified by PLC (hexane-EtOAc, 1:1). The upper band, though homogeneous on TLC in several solvent systems, was a 3:1 mixture of 2g and 2h by NMR spectrometry, mp 155-157° (hexane-CHCl₃), yield 0.10 g, $[\alpha]_D - 375^\circ$ (c 0.016, CHCl₃), IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3415, 1770, 1730, 1720, 1695, 1662, 1630, 1625 and 1615. (Calc. for 2g, C23H28O8: MW, 432.1783. Found: MW(MS), 432.1803, 0.6%). M⁺ of 2h was not observed. Other significant peaks in the high resolution MS were at m/e (rel. int.): 414 ($C_{23}H_{26}O_7$, 1.1), $347 (C_{19}H_{23}O_6, 1.9), 346 (C_{19}H_{22}O_6, 4.6), 278 (C_{15}H_{18}O_5,$ 4.8), 261 ($C_{15}H_{17}O_4$, 260 ($C_{15}H_{16}O_4$, 11), 233 ($C_{14}H_{17}O_3$, 4.3), 232 $(C_{14}H_{16}O_3, 6.8)$, 217 $(C_{13}H_{13}O_3, 11.7)$, 125 $(C_7H_9O_2, 10.2), 85 (C_6H_{13}O, 51), 83 (C_6H_{11}O, 47.2), 82$ $(C_5H_6O, 100)$, 71 (C_4H_7O) and 69 $(C_4H_5O, low resoln.)$. The lower band gave a 5:1 mixture of 4a and 4b, mp 105-108° (CHCl₃-hexane, yield 0.15 g, $[\alpha]_D$ - 192° (c 0.089, CHCl₃); CD curve $[\theta]_{290}$ – 4600, $[\theta]_{250}$ + 4550, $[\theta]_{214}$ – 14 700 (last reading); IR $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$: 3410, 1775, 1730, 1720, 1708, 1660, 1630, 1620, 910 and 820. (Calc. for 4a, C₂₄H₃₀O₉; MW, 462.1939: Found: MW(MS), 462.1905, 9.1%). M+ of 4b was not observed. Other significant peaks in the high resolution MS were at m/e (rel. int.): 444 ($C_{24}H_{28}O_8$, 3.4), 376 $(C_{20}H_{24}O_7, 4.8)$, 224 $(C_{12}H_{16}O_4, 9.2, 206)$ $(C_{12}H_{14}O_3, 9.2, 206)$ 8.8), 168 ($C_9H_{12}O_3$, 4.4), 164 ($C_{10}H_{12}O_2$, 6.9), $C_{10}H_{11}O_2$, 4.7), 153 ($C_8H_9O_3$, 3.2), 151 ($C_9H_{11}O_2$, 4.4), 125 ($C_7H_9O_2$, 4.0). 85 (C₅H₉O, 57.5), 83 (C₅H₇O, 100), 82 (C₅H₆O, 74.5), 69 (C₄H₅O, low resoln.). Fractions 21–26 were further purified by PLC (hexane-EtOAc 1:1, 3 developments). The more mobile fraction was a 2:1 mixture of 4c and 4d, mp 120-122°, wt 0.32 g, $[\alpha]_D - 169^\circ$ (c 0.098, CHCl₃); CD curve (MeOH), $[\theta]_{295} - 4500$, $[\theta]_{248} + 4450$, $[\theta]_{214} - 33750$ (last reading); IR $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$: 3420, 1770, 1730, 1720, 1660, 1630, 920 and 820. (Calc. for 4c, C23H28O9: MW, 448.1782. Found: MW(MS), 448.1752, 2.5%). M+ of 4d was not observed. Other significant peaks in the high resolution MS were at m/e (rel. int.): 420 ($C_{22}H_{28}O_8$, 3.3), 363 ($C_{19}H_{23}O_7$, 2.2), 362 ($C_{19}H_{22}O_7$, 2.0), 319 ($C_{17}H_{19}O_6$, 2.4), 292 $(C_{16}H_{20}O_5, 21.0), 277 (C_{15}H_{17}O_5, 2.4), 276 (C_{15}H_{16}O_5,$ 6.8), 249 $(C_{14}H_{17}O_4, 3.6)$, 248 $(C_{14}H_{16}O_8, 2.9)$, 224 $(C_{12}H_{16}O_4, 21.2), 206 (C_{12}H_{14}O_3, 49.5), 205 (C_{12}H_{13}O_3,$ 11.2), 193 ($C_{11}H_{13}O_3$, 8.2), 178 ($C_{11}H_{14}O_2$, 21.8), 164 $(C_{10}H_{12}O_2, 23.3), 149 (C_8H_5O_3, 23.7), 83 (C_5H_7O, 76.6), 82$ $(C_5H_6O, 28.7)$, 71 (C_4H_7O) and 69 (C_4H_5O) , low resoln.). The

less mobile fraction was non-crystalline **4e**, yield $0.18~g~[\alpha]_D-106^\circ~(c~0.044,~CHCl_3);~CD~curve~(MeOH)~[\theta]_{292}-4200, [\theta]_{245}+4200, [\theta]_{217}-42~200~(last~reading);~IR~\nu_{max}^{KBr}~cm^{-1}:3420,~1780,~1750,~1725,~1708,~1650~and~1235.~(Calc.~for~C_{21}H_{26}O_9;~MW,~422.1575.~Found:~MW(MS)~422.1575).$ Other significant peaks in the high resolution MS were at $\emph{m/e}$ (rel. int.): 295 $(C_{15}H_{19}O_6,~1.0),~294~(C_{15}H_{18}O_6,~3.8),~276~(C_{15}H_{16}O_5,~8.1),~266~(C_{14}H_{18}O_5,~8.2),~250~(C_{13}H_{14}O_5,~9.3),~224~(C_{12}H_{16}O_4,~62.6),~206~(C_{12}H_{14}O_3,~31.8),~164~(C_{10}H_{12}O_2,~27.3).~153~(C_8H_9O_3,~14.3),~151~(C_9H_{11}O_2,~16.4),~136~(C_8H_8O_2,~12.0),~125~(C_7H_9O_2,~18.2),~97~(C_6H_9O,~27.6),~83~(C_5H_7O,~23),~82~(C_5H_6O,~23.5)~and~69~(C_4H_5O,~low resoln.).$

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